

Rehydration of Whey Protein Isolate: Effect of Temperature, Water Activity, and Storage Time

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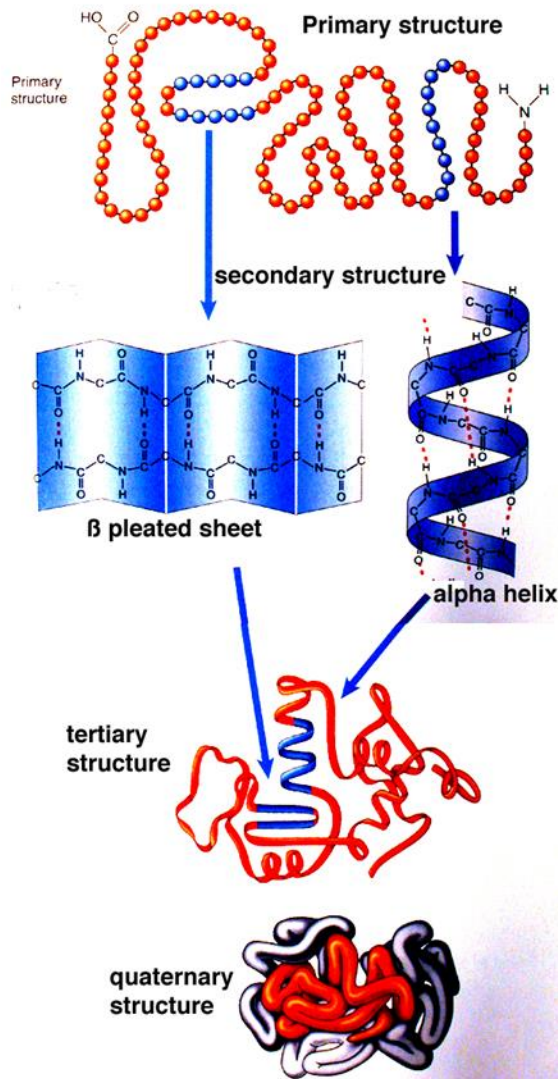
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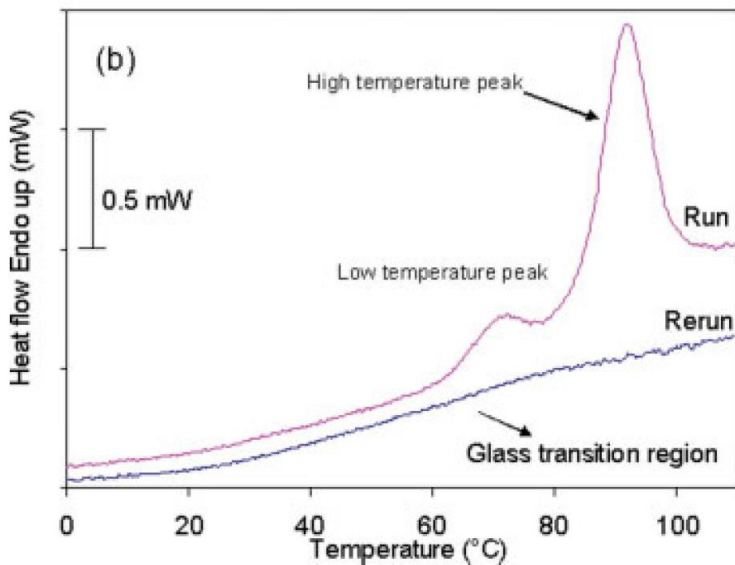
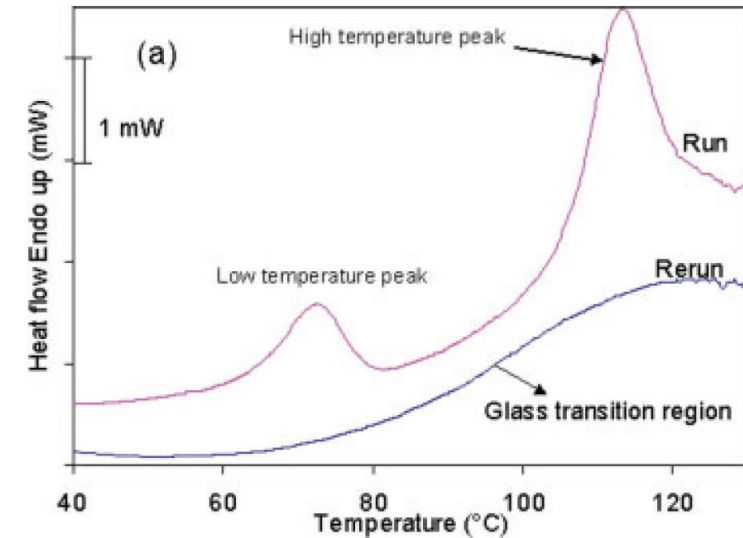
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Background



- Hydration properties of whey protein powders are poorly understood
- Conformational changes occur at ambient conditions during storage
- Such structural changes may increase the hydrophilic area resulting in a decrease water activity of WPI
- Various physico-chemical properties are dependent on temperature and water content

Thermal Properties of Proteins



- Biological polymers often show a low temperature endotherm that appears below glass transition
- Farahnaky et al. (2005) interpreted such endotherm in bovine serum albumin as enthalpy relaxation
- Potes and Roos (2015) explained such time-dependent endotherm as hydration/dehydration of protein.

Objectives

- To understand the hydrodynamics of water-protein behavior on the reversible endothermic transition
- To understand the effect of storage time and temperature on WPI conformational structure

**Whey protein isolate
(WPI)**



Experimental

Differential Scanning Calorimetry (DSC)

WPI powder with various a_w was prepared using saturated salt solutions in humidified vacuum sealed desiccators at RT

WPI samples with different a_w
Heat-scanning: $-20^{\circ}\text{C} - 50^{\circ}\text{C}$,
 $5^{\circ}\text{C}/\text{min}$ with 2 cycles.
Repeated after storing at RT
for 7d.

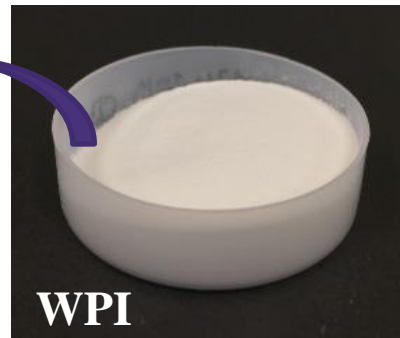
Vacuum Desiccator



Experimental

Water Activity Measurement

WPI samples with various a_w .
Temperature range 20°C-50°C
at 5°C intervals for 3 cycles of
heating-cooling.
Repeat after storing for 7d
& 14 d at RT



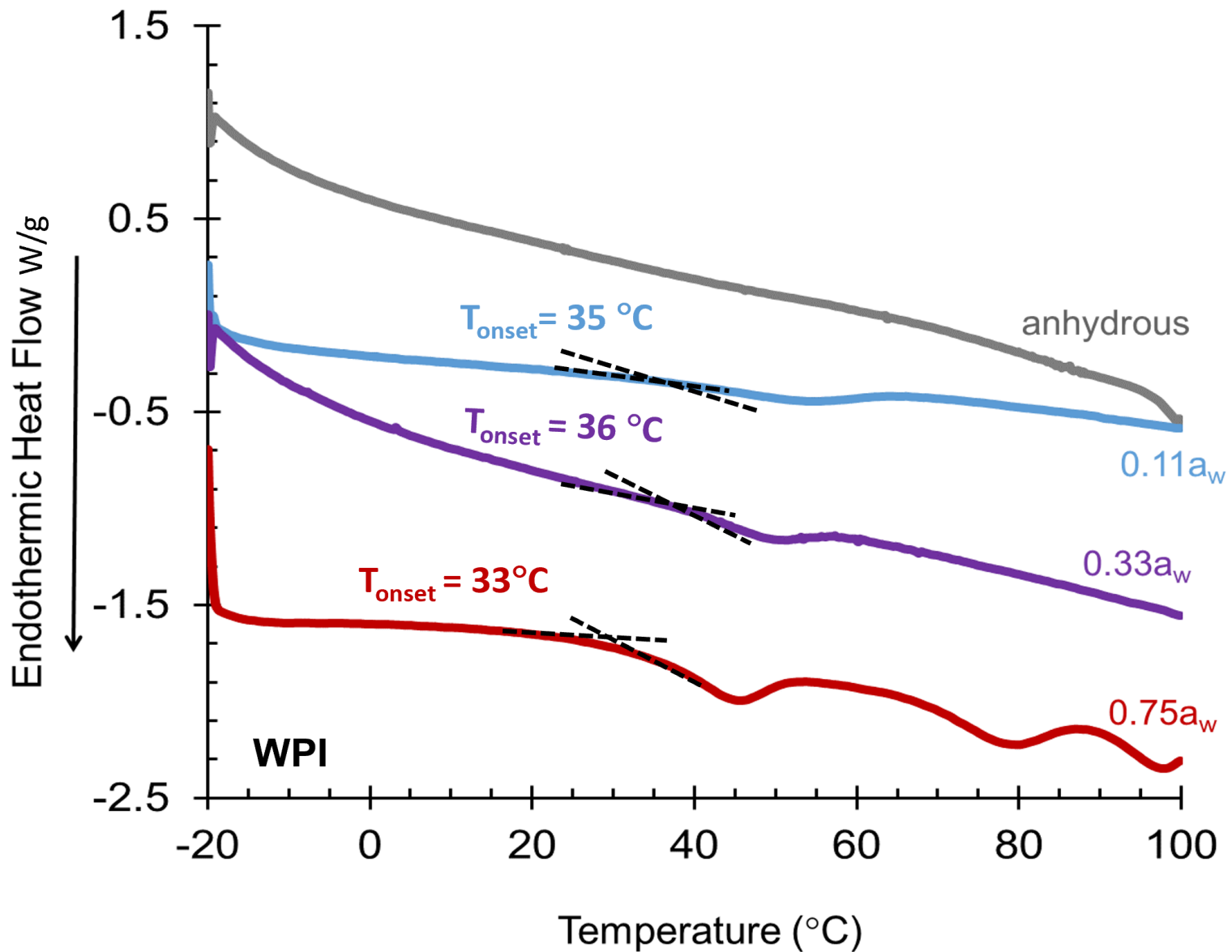
Apparent Viscosity

Sample: WPI powder
dispersed in water (5% -
40%, w/w), mixed for 1
h at 5°C -50°C.
Shear rate: 250 s⁻¹

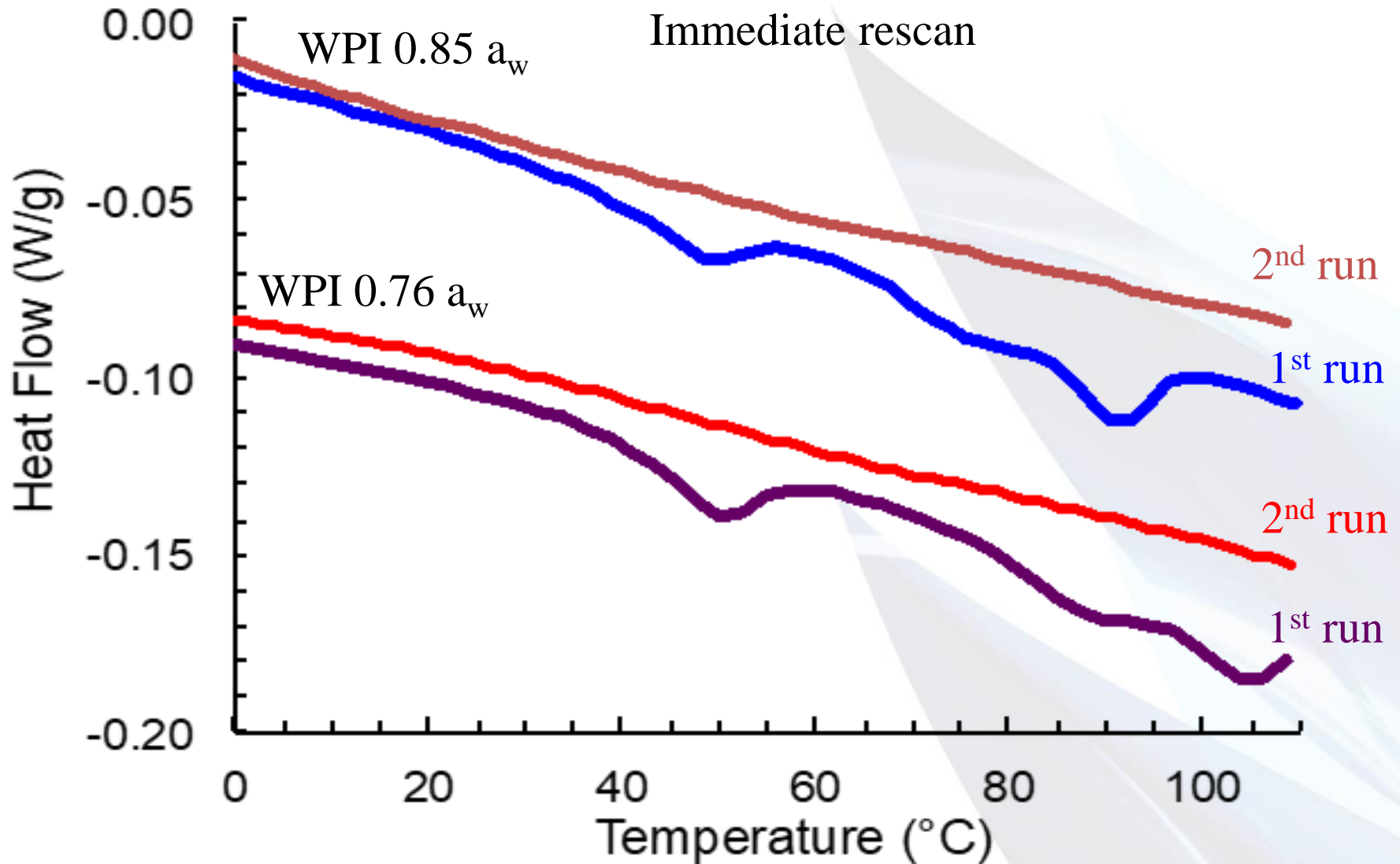


Results and Discussions

Thermal analysis (DSC)



Thermal analysis (DSC)

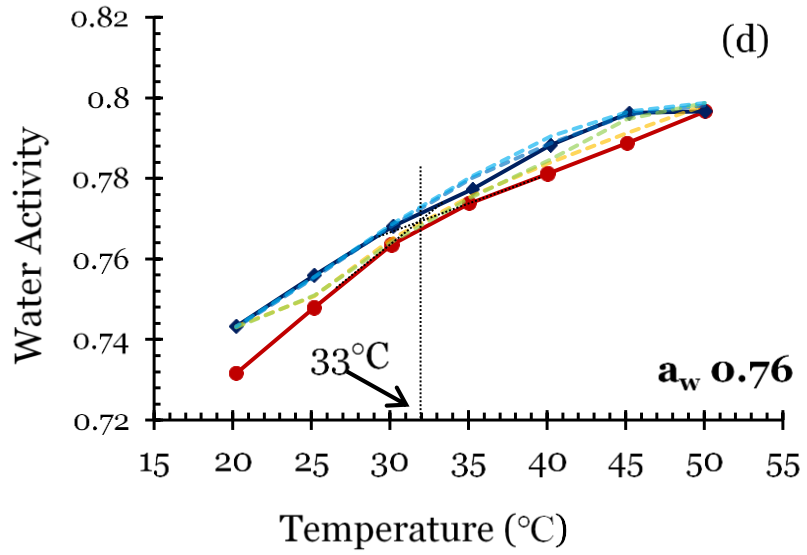
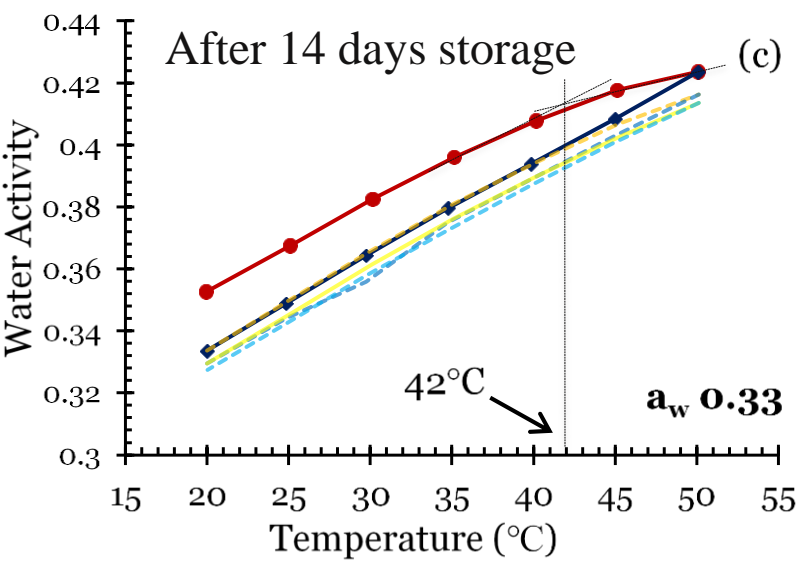
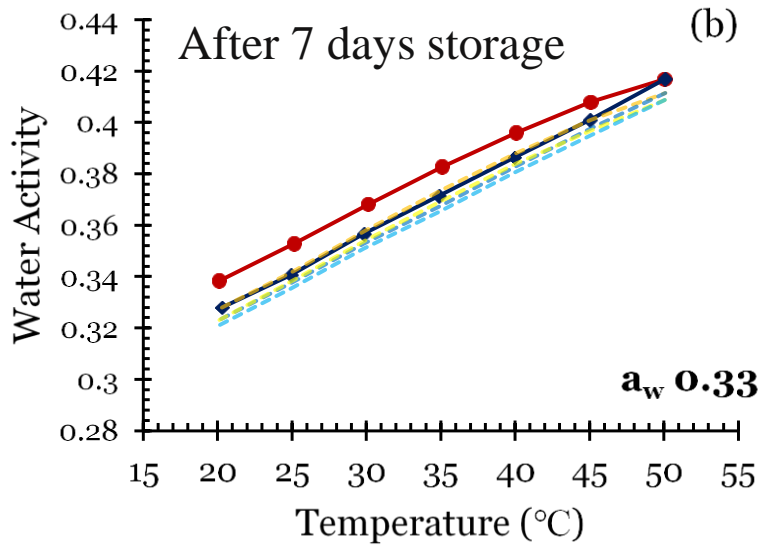
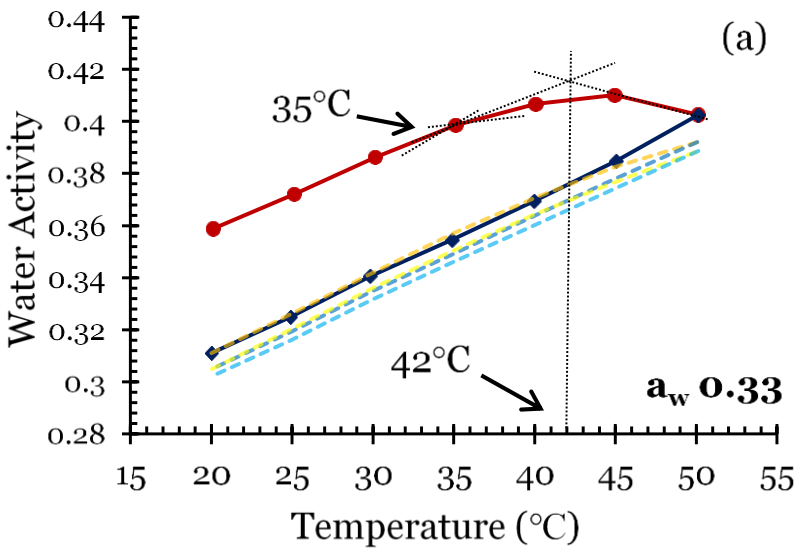


Thermal analysis (DSC)

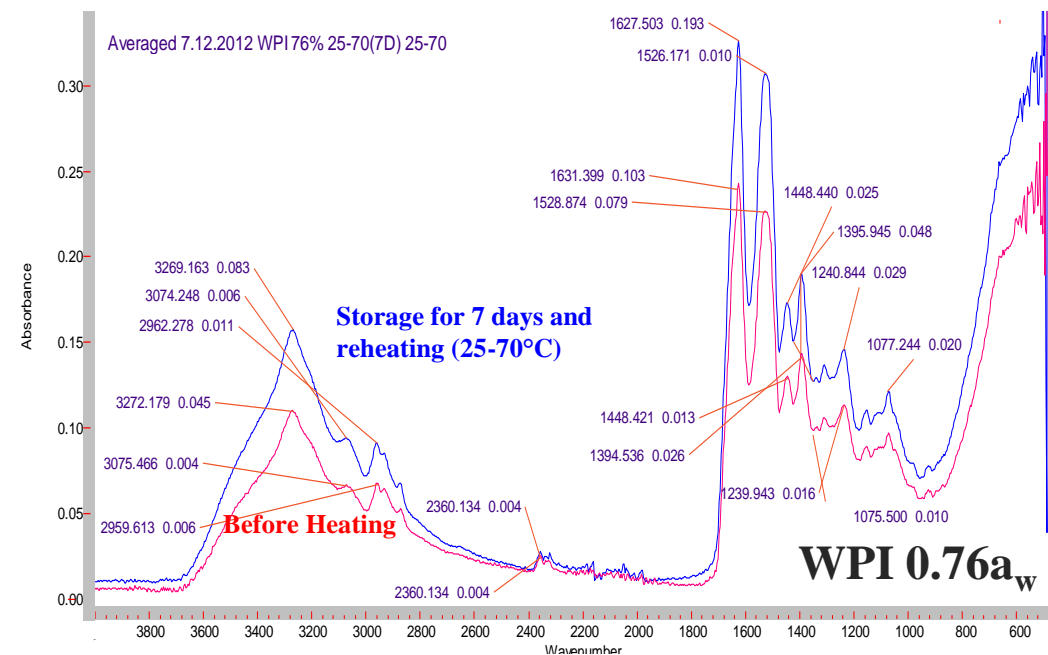
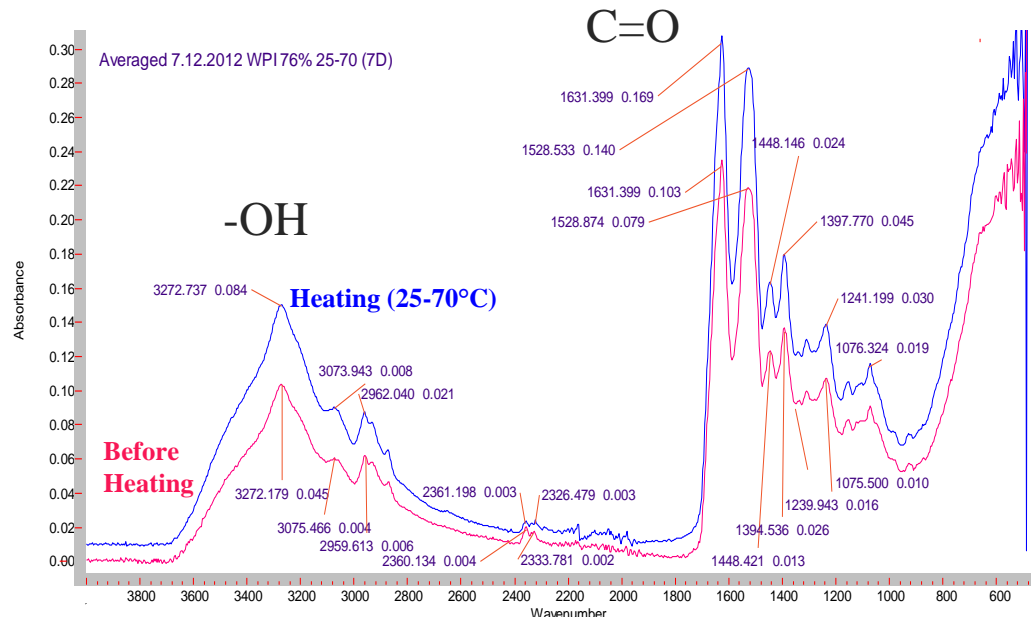
WPI scanned up to 50 °C, and again after 7 days storage

RH%	First Scan				Rescan after stored for 7 Days			
	1 st heating		2 nd heating		1 st heating		2 nd heating	
	T _{onset} (°C)	T _{endo} (°C)	T _{onset} (°C)	T _{endo} (°C)	T _{onset} (°C)	T _{endo} (°C)	T _{onset} (°C)	T _{endo} (°C)
11	23	30	-	-	-	-	-	-
23	-	-	-	-	-	-	-	-
33	-	-	-	-	-	-	-	-
44	29	40	-	-	-	-	-	-
54	32	41	-	-	34	45	-	-
65	28	39	-	-	32	44	-	-
76	30	41	-	-	32	43	-	-
85	28	39	-	-	30	42	-	-

Water Activity Measurement of WPI



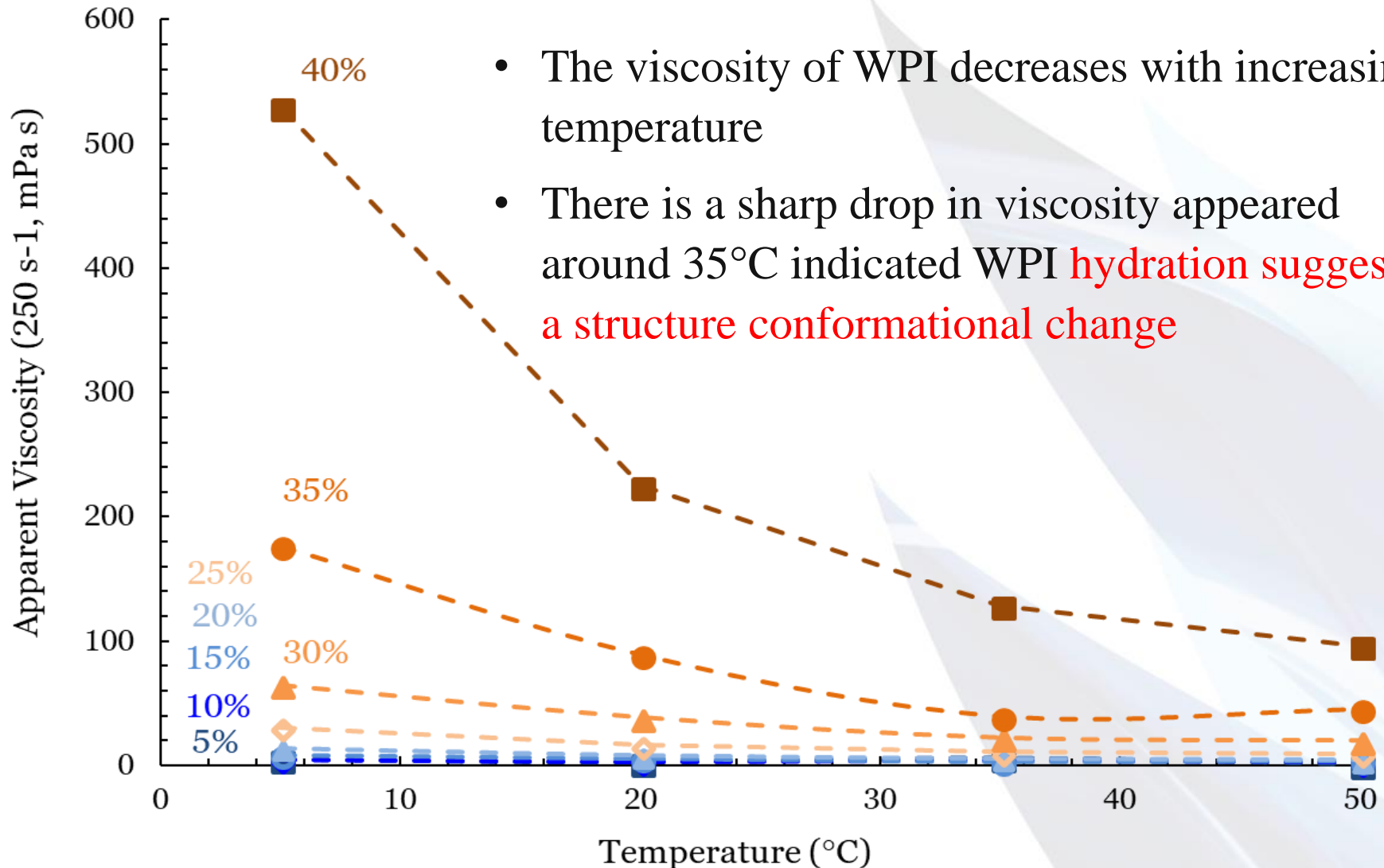
—●— Heating
 —◆— Cooling
 - - - Subsequent cooling-heating



- FTIR spectra are similar before and after heating followed by 7 days of storage
- Reheating up to 70°C shifts FTIR absorbance peaks at several wavelengths
- Largest shifts found for –OH groups at 2500 to 3300 nm⁻¹

Apparent Viscosity (WPI Dispersion)

Apparent Viscosity of WPI aqueous dispersions at a fix shear rate 250s^{-1} as a function of temperature



- The viscosity of WPI decreases with increasing temperature
- There is a sharp drop in viscosity appeared around 35°C indicated WPI hydration suggesting a structure conformational change

Conclusion

- A water content and time-dependent reversible endotherm occurs in WPI within 30°C - 40°C
- Heating WPI from 25°C - 50°C results in increased protein hydration according to a consequent decrease and hysteresis of a_w during re-cooling
- FTIR data show that the time-dependent endotherm is related to changes in hydrogen bonding (hydration/dehydration)
- Viscosity decreases significantly at $\sim 35^\circ\text{C}$ as a function of WPI concentration and temperature

Acknowledgment:



UCC

University College Cork, Ireland
Coláiste na hOllscoile Corcaigh



مؤسسة الكويت للتقدم العلمي
Kuwait Foundation for the Advancement of Sciences

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